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Two new isoflavone glycosides from *Pueraria lobata*

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Two new isoflavone diglycosides, formononetin 8-*C*-[β -D-apiofuranosyl-(1 \rightarrow 6)]- β -D-glucopyranoside (**1**) and formononetin 8-*C*-[β -D-xylopyranosyl-(1 \rightarrow 6)]- β -D-glucopyranoside (**2**), were isolated from the roots of *Pueraria lobata*, together with four known compounds, 4'-methoxypterarin (**3**), daidzin (**4**), genistin (**5**), and daidzein (**6**). The structures of these compounds were elucidated by the spectroscopic methods.

Keywords: *Pueraria lobata*; Leguminosae; isoflavone glycosides; formononetin 8-*C*-[β -D-apiofuranosyl-(1 \rightarrow 6)]- β -D-glucopyranoside; formononetin 8-*C*-[β -D-xylopyranosyl-(1 \rightarrow 6)]- β -D-glucopyranoside

1. Introduction

Pueraria lobata (Willd.) ohwi (Leguminosae) is a perennial plant native to eastern Asia. Its root is an important Chinese traditional medicine as an antipyretic, antidiarrhetic, diaphoretic, and antiemetic agent [1–4]. Previous phytochemical investigations on this plant have led to the isolation of various isoflavones [5], aromatic glycosides [6], and saponins [7–9]. In our recent phytochemical research for this plant, two new isoflavone glycosides, formononetin 8-*C*-[β -D-apiofuranosyl-(1 \rightarrow 6)]- β -D-glucopyranoside (**1**) and formononetin 8-*C*-[β -D-xylopyranosyl-(1 \rightarrow 6)]- β -D-glucopyranoside (**2**), were isolated along with four known isoflavonoid glycosides, 4'-methoxypterarin (**3**), daidzin (**4**), genistin (**5**), and daidzein (**6**). Their structures were determined by the MS and NMR spectroscopic analysis.

2. Results and discussion

The roots of *P. lobata* were extracted with boiling water, and the extract was precipitated with ethanol. The ethanol-soluble part was subjected to column chromatography (CC) with X-5 macroporous resin, industrial HPLC, and polyamide to furnish two new isoflavone diglycosides (**1** and **2**) and four known isoflavones (**3–6**).

Compound **1** was obtained as white amorphous powder. The molecular formula, C₂₇H₃₀O₁₃, was confirmed on the basis of the [M – H][–] ion at *m/z* 561.1632 in the HR-ESI-MS spectra. The IR spectrum of **1** revealed the presence of hydroxyl group (3419 cm^{–1}), carbonyl group (1631 cm^{–1}), and aromatic ring (1608, 1588, and 1514 cm^{–1}). The absorption maxima at 250 and 304 nm in the UV spectrum and the proton singlet at δ 8.38 (H-2) in the ¹H NMR spectrum suggested **1** to be an isoflavone.

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Table 1. ^1H NMR and ^{13}C NMR spectral data for compounds **1** and **2** ($\text{DMSO-}d_6$)^{a,b}.

No.	1		2	
	^1H	^{13}C	^1H	^{13}C
2	8.38 s	152.8	8.40 s	152.9
3		124.2		124.1
4		174.7		174.7
5	7.94 d (8.5)	126.1	7.95 d (9.0)	126.2
6	7.00 d (8.5)	113.9	7.00 d (9.0)	114.0
7		161.5		161.0
8		112.4		112.5
4a		116.4		116.5
8a		156.1		156.2
1'		122.6		122.7
2'	7.53 d (9.0)	129.9	7.53 d (9.0)	129.9
3'	6.99 d (9.0)	113.5	6.99 d (9.0)	113.5
4'		158.8		158.8
5'	6.99 d (9.0)	113.5	6.99 d (9.0)	113.5
6'	7.53 d (9.0)	129.9	7.53 d (9.0)	129.9
1''	4.81 d (9.5)	73.3	4.79 d (9.0)	73.2
2''	4.04 dd (9.5, 9.0)	70.5	4.01 dd (9.0, 10.0)	69.8
3''	3.27 m	78.7	3.26 m	78.5
4''	3.19 m	70.4	3.20 m	70.3
5''	3.40 d (8.0)	79.9	3.40 m	80.0
6''	a: 3.94 m b: 3.39 d (8.0)	68.3	a: 4.02 m b: 3.26 m	69.2
1'''	4.78 d (3.0)	108.9	4.10 d (7.5)	103.8
2'''	3.74 d (3.0)	75.5	2.93 m	73.1
3'''		78.6	3.04 m	76.4
4'''	a: 3.84 m b: 3.57 d (9.5)	73.1	3.53 m	69.4
5'''	3.31 (m)	62.8	a: 3.68 (m) b: 3.00 t (11.0)	65.5
4'-OCH ₃	3.79 s	55.0	3.79 s	55.0

^a Signals were assigned by means of the HMQC and HMBC spectra (J Hz).

^b Experiments were done on 500 MHz for the ^1H and 125 MHz for ^{13}C NMR spectra, respectively, and the J values (parentheses) are in Hertz.

its anomeric proton due to axial interaction. According to the above evidence, **1** was characterized as formononetin 8- C -[β -D-apiofuranosyl-(1 \rightarrow 6)]- β -D-glucopyranoside (Figure 1). The ^1H NMR and ^{13}C NMR spectral data were assigned by the HMBC and HMQC spectra and listed in Table 1.

Compound **2** was obtained as white amorphous powder. The molecular formula, $\text{C}_{27}\text{H}_{30}\text{O}_{13}$, was established by negative HR-ESI-MS at m/z 561.1624 [$\text{M} - \text{H}$]⁻. Its UV absorption maxima, the negative ESI-MS, and ESI-MS/MS spectra were completely identical to those of compound **1**, suggesting the similar aglycone and sugar moieties in **1** and **2**. The ^1H NMR, ^{13}C NMR spectra (Table 1) and the DEPT-135 spectral data of **2**

showed close resemblance with those of **1**, except for the signals due to terminal sugar unit. The nature of the terminal sugar units of **2** was evident from the ^{13}C NMR spectrum that showed the presence of one O -xylopyranosyl unit, and the terminal sugar unit was confirmed to be linked to the C-6'' of the glucose on the basis of the correlations between the proton at δ 4.10 (xyl H-1) and the carbons at δ 69.2 (Glu C-6), 73.1 (xyl C-2), and 76.4 (xyl C-3) in the HMBC spectrum. These data are consistent with those of daidzein 8- C -[β -D-xylopyranosyl-(1 \rightarrow 6)]- β -D-glucopyranoside [5]. The β -configuration of the two sugars was confirmed by the J values ($J = 9.0$ and 7.5 Hz) of their anomeric protons. Thus, **2** was identified as formononetin

8-C-[β -D-xylopyranosyl-(1 \rightarrow 6)]- β -D-glucopyranoside (Figure 1).

Compound **3** was determined as 4'-methoxypterarin by comparison of the ^1H NMR, ^{13}C NMR, IR, UV, and MS spectral data with those of compounds **1** and **2**. The structures of the other three known compounds were identified as daidzin (**4**), genistin (**5**), and daidzein (**6**) by comparison of the ^1H NMR, ^{13}C NMR, IR, UV, and MS spectral data with the values in literature [5].

3. Experimental

3.1 General experimental procedures

Melting points were obtained using an XT-4 melting point apparatus and are uncorrected. Optical rotations were measured on a JASCO P-1020 polarimeter. The UV and IR spectra were recorded on a Waters 2996 photodiode array detector (PDA) and a Perkin-Elmer 683 infrared spectrophotometer in KBr disks, respectively. The high-resolution mass spectra were recorded on the Macromass Q-TOF mass spectrometer fitted with a Lockspray interface. The MS/MS experiments were performed on a Finnigan TSQ/SSQ triple quadrupole mass spectrometer equipped with an ESI interface. Argon was used as the collision gas and the pressure was set at 3.0 mTorr. The ^1H NMR and ^{13}C NMR spectra were recorded on Bruker DRX400 (400 MHz for ^1H and 100 MHz for ^{13}C) and Bruker AV500 (500 MHz for ^1H and 125 MHz for ^{13}C) spectrometer. Analytical HPLC experiments used to monitor each fraction were carried out on Waters 2695 chromatograph with a PDA detector and a Tigerkin ODS-2 column (5 μm , 4.6 \times 250 mm I.D., Dalian Sipore Co. Ltd, Dalian, Liaoning, China). CC was performed on X-5 macroporous resin and polyamide with a pump. Preparative HPLC was carried out on self-made industrial chromatography with a single wavelength ultraviolet detector (310 nm) and a Tigerkin ODS-2 column (10–20 μm , 80 \times 360 mm, I.D.), and the flow was set at 200 ml/min.

3.2 Plant material

The dried roots of *P. lobata* were collected in Yuexi, Anhui Province of China, in November 2004 and authenticated by Xirong He, Institute of Medication, Xiyuan Hospital of China Academy of Traditional Chinese Medicine. A voucher specimen (0471) is deposited in the Dalian Institute of Chemical Physics, Chinese Academy of Science, China.

3.3 Extraction and isolation

The dried roots of *P. lobata* (10 kg) were extracted with boiling water (100 L \times 2) under reflux for 2 h, and the combined extracts were concentrated. The residue (2.0 kg) was dissolved in H_2O and then precipitated with 65 and 80% ethanol, respectively. After removal of the deposition, the soluble part was concentrated *in vacuo* to yield 968 g of the residue. The residue was subjected to CC on the X-5 macroporous resin (30 L), eluted with water, and then 10, 20, 40, 70, and 95% EtOH to get six fractions (A–F). In each eluting condition, 120 L was used. Fraction F (15.9 g) was subjected to preparative HPLC with MeOH (A)– H_2O (B) as mobile phase at a flow of 200 ml/min. The gradient profile was as follows: 0–45 min linearly from 30% A to 55% A, 45–50 min linearly from 55% A to 70% A, and then held for 10 min. Compounds **4** (78 mg) and **6** (40 mg), as well as 9 fractions (F-1–F-9), were also obtained. Fraction F-4 (0.8 g, $t_{\text{R}} = 30.4$ –36.0 min) was subjected to CC on the polyamide (950 ml) and eluted with water, and then 15, 30, 50, and 95% EtOH. In each eluting condition, 3800 ml was used and the samples collected by each 950 ml. Compounds **1** (85 mg), **2** (21 mg), **3** (16 mg), and **5** (45 mg), as well as other 13 fractions, were obtained.

3.3.1 Formononetin 8-C-[β -D-apiofuranosyl-(1 \rightarrow 6)]- β -D-glucopyranoside (**1**)

White amorphous powder (H_2O), mp 166–168°C; $[\alpha]_{\text{D}}^{25}$: 27.3 (*c* 0.82, DMSO); UV λ_{max} (ACN/ H_2O) nm: 250, 304; IR (KBr) ν_{max} :

3419 (OH), 2931 (CH), 1631 (C=O), 1610 (aromatic), 1588, 1514, 1445, 1249, 1180, and 1082 cm^{-1} ; ^1H NMR and ^{13}C NMR spectral data: see Table 1; HR-ESI-MS m/z 561.1632 $[\text{M} - \text{H}]^-$ (calcd for $\text{C}_{27}\text{H}_{30}\text{O}_{13}$, 561.1608); positive ion ESI-MS m/z : 563.1 $[\text{M} + \text{H}]^+$, 585.1 $[\text{M} + \text{Na}]^+$; negative ion ESI-MS m/z : 561.1 $[\text{M} - \text{H}]^-$; negative ion ESI-MS/MS m/z : 429.3 $[\text{M} - \text{H} - \text{Api}]^-$, 375.4 $[\text{M} - \text{H} - \text{Api} - 3\text{H}_2\text{O}]^-$, 339.4 $[\text{M} - \text{H} - \text{Api} - \text{C}_3\text{H}_6\text{O}_3]^-$, 309.4 $[\text{M} - \text{H} - \text{Api} - \text{C}_4\text{H}_8\text{O}_4]^-$, 281.3 $[\text{M} - \text{H} - \text{Api} - \text{C}_4\text{H}_8\text{O}_4\text{-CO}]^-$.

3.3.2 Formononetin 8-C- $[\beta\text{-D-xylopyranosyl-}(1 \rightarrow 6)]\text{-}\beta\text{-D-glucopyranoside}$ (2)

White amorphous powder (H_2O), mp 143–145°C; $[\alpha]_{\text{D}}^{25}$: -9.2 (c 0.55, DMSO); UV λ_{max} (ACN/ H_2O) nm: 250, 307; IR (KBr) ν_{max} : 3427 (OH), 2918 (CH), 1633 (C=O), 1608 (aromatic), 1589, 1514, 1331, 1249, 1211, and 1113 cm^{-1} ; ^1H NMR and ^{13}C NMR spectral data: see Table 1; HR-ESI-MS m/z : 561.1624 $[\text{M} - \text{H}]^-$ (calcd for $\text{C}_{27}\text{H}_{30}\text{O}_{13}$, 561.1608); positive ion ESI-MS m/z : 563.1 $[\text{M} + \text{H}]^+$, 585.1 $[\text{M} + \text{Na}]^+$; negative ion ESI-MS m/z : 561.1 $[\text{M} - \text{H}]^-$; negative ion ESI-MS/MS m/z : 429.3 $[\text{M} - \text{H} - \text{Xyl}]^-$, 375.4 $[\text{M} - \text{H} - \text{Api} - 3\text{H}_2\text{O}]^-$, 339.4 $[\text{M} - \text{H} - \text{Xyl} - \text{C}_3\text{H}_6\text{O}_3]^-$, 309.4 $[\text{M} - \text{H} - \text{Api} - \text{C}_4\text{H}_8\text{O}_4]^-$, $[\text{M} - \text{H} - \text{Api} - \text{C}_4\text{H}_8\text{O}_4\text{-CO}]^-$.

3.3.3 4'-Methoxypuerarin (3)

White amorphous powder ($\text{MeOH} - \text{H}_2\text{O}$), mp 151–153°C; $[\alpha]_{\text{D}}^{25}$: $+10.2$ (c 0.54, DMSO); UV λ_{max} (ACN/ H_2O) nm: 250, 304; ^1H NMR (400 MHz, DMSO- d_6): δ 8.38 (1H, s, H-2), 7.97 (1H, d, $J = 8.8$ Hz, H-5), 7.02 (1H, d, $J = 8.8$ Hz, H-6), 7.52 (2H, d, $J = 8.8$ Hz, H-2',6'), 7.00 (2H, d, $J = 8.8$ Hz, H-3',5'), 4.84 (1H, d, $J = 8.8$ Hz, Glu H-1), 3.79 (1H, s, 4'-OCH₃); ^{13}C NMR (125 MHz, DMSO- d_6): δ 153.2 (C-2), 124.4 (C-3), 175.2 (C-4), 126.6 (C-5), 113.9 (C-6), 161.3 (C-7), 112.8 (C-8),

117.0 (C-4a), 156.3 (C-8a), 123.0 (C-1'), 130.3 (C-2',6'), 115.0 (C-3',5'), 159.2 (C-4'), 73.6 (Glu C-1), 70.9 (Glu C-2), 78.7 (Glu C-3), 70.6 (Glu C-4), 81.9 (Glu C-5), 61.5 (Glu C-6), 55.4 (4'-OCH₃); positive ion ESI-MS m/z 431.0 $[\text{M} + \text{H}]^+$; negative ion ESI-MS m/z 429.0 $[\text{M} - \text{H}]^-$.

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References

- [1] W.M. Keung and B.L. Vallee, *Phytochemistry* **47**, 499 (1998).
- [2] W.M. Keung and B.L. Vallee, *Proc. Natl Acad. Sci.* **90**, 10008 (1993).
- [3] S.M. Boue, T.E. Wiese, S. Nehls, S. Elliott, C.H. Carter-Wientjes, B.Y. Shih, J.A. McLachlan, and T.E. Cleveland, *J. Agric. Food Chem.* **51**, 2193 (2003).
- [4] J.S. Lee, *Clin. Chim. Acta* **347**, 121 (2004).
- [5] K. Junei, F. Junich and C. Junko, *Chem. Pharm. Bull.* **35**, 4846 (1987).
- [6] N. Toshihiro, K. Junei, F. Junichi, S. Yusuke, I. Mutsumi, S. Yoshiaki, I. Yayoi, Y. Ichiro, and K. Manki, *Phytochemistry* **33**, 1207 (1993).
- [7] T. Arao, J. Kinjo, T. Nohara, and R. Isobe, *Chem. Pharm. Bull.* **45**, 362 (1997).
- [8] T. Arao, T. Idzu, J. Kinjo, T. Nohara, and R. Isobe, *Chem. Pharm. Bull.* **44**, 1970 (1996).
- [9] T. Arao, J. Kinjo, T. Nohara, and R. Isobe, *Chem. Pharm. Bull.* **43**, 1176 (1995).
- [10] G. Rath, A. Touré, M. Nianga, J.L. Wolfender, and K. Hostettmann, *Chromatographia* **41**, 332 (1995).
- [11] P. Waridel, J.L. Wolfender, K. Ndjoko, K.R. Hobby, H.J. Major, and K. Hostettmann, *J. Chromatogr. A* **926**, 29 (2001).
- [12] C.C.W. Wanjala and R.R.T. Majinda, *Phytochemistry* **51**, 705 (1999).
- [13] J. Cheng, Y.Y. Zhao, B. Wang, L. Qiao, and H. Liang, *Chem. Pharm. Bull.* **53**, 419 (2005).